

Chapter Test B Chemical Reactions Answers

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Chapter Test B Chemical Reactions

Chapter Test B, continued ____ 7. In a reaction, the ions of two compounds exchange places in aqueous solution to form two new compounds. This reaction is called a a. synthesis reaction. b. decomposition reaction. c. single-displacement reaction. d. double-displacement reaction. ____ 8. The use of a double arrow in a chemical equation indicates that the reaction a. is reversible.

Assessment Chapter Test B - Ed W. Clark High School

A chemical reaction in which energy is absorbed is called a. synthetic. b. exothermic. c. endothermic. ____ 4. Chemical equations must be balanced because chemical reactions obey the principle of a. activation. b. conservation of matter. c. decomposition. ____ 5. In a chemical equation, the number in front of a chemical

sxPS G8 CT ch06-B 067-070

b) only reactions in which a catalyst is present can reach chemical equilibrium. c) if a chemical equilibrium is disturbed, the system will try and reestablish equilibrium. d) if a chemical equilibrium is disturbed, the temperature of the system will always decrease.

CHAPTER 9

Imagine that the soccer ball and the soccer goal are reactants. The reaction you want to take place will combine the soccer ball and the goal to score a point. The combination of the goal and the ball into something called a point is the product. However, the ball and the goal start meters away from each other.

Chemical Reaction; Grade 8 Chapter 8

CHAPTER 8 REVIEW Class SĒc7ĪB-n Chemical Equations and Reactions SHORT ANSWER Answer the following questions in the space provided. 1. Match the symbol on the left with its appropriate description on the right. (aq) (a) (b) (c) (d) (f) A precipitate forms. A gas forms. A reversible reaction occurs. Heat is applied to the reactants.

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The only sure evidence for a chemical reaction is a. the formation of a gas. b. changes in properties. When a chemical reaction takes place in an open system, a. matter cannot move at all. b. matter can enter from or escape to the surroundings.

Chapter 2 Chemical Reactions Practice Test - MAFIADOC.COM

When magnesium is burnt in air then (i) Magnesium is reacting with oxygen (ii) Magnesium is reacting with nitrogen (iii) Magnesium is reacting with carbon (iv) Magnesium is reacting with Carbon di oxide Q9. Write values of a,b and c if following chemical reaction is balanced .

Chapter 1 Chemical Reactions and Equations MCQ Test 1 ...

Chemical Reactions - Chapter 19. Terms in this set (20) Balanced Chemical Equation. chemical equation with the same number of atoms of each element on both sides of an equation. catalyst. substance that speeds up a chemical reaction without being permanently changed itself. chemical equation.

Chemical Reactions - Chapter 19 Flashcards | Quizlet

a substance that increases the rate of reaction by lowering the activation energy barrier; the catalyst is not used up in reactions what is a spectator ion an ion that is not directly involved in a chemical reaction, an ion that doesn't change oxidation number or composition during a reaction

Chemistry chapter 11 test Flashcards | Quizlet

Question 12.Represent each of the following word equations with a balanced chemical equation. a)Disilane gas (Si 2 H 6) undergoes combustion to form solid silicon dioxide and water. b)Solid aluminum hydride is formed by a combination reaction of its two elements.

Chemical reactions and equations class 10 test

Chapter 7 TEST CHEMICAL REACTIONS PART 1: MULTIPLE CHOICE. ____ 1. Which of these is not a sign of a chemical reaction? a. A gas is given off. c. Heat is released. b. The material dissolves. d. A color change occurs. ____ 2. The substance that is formed in a chemical reaction is called the

Chapter 7 TEST CHEMICAL REACTIONS

Chemical Reactions 39 Chapter Test B Chemical Reactions I. Testing Concepts Directions: For each of the following, write the letter of the term or phrase that best completes the sentence or answers the question. 1. Substances that prevent certain chemical reactions are called ____ a. endothermics. b. catalysts. c. inhibitors. d. products. 2.

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Chapter 7 Pretest: Chemical Reactions Multiple Choice Identify the choice that best completes the statement or answers the question. ____ 1. When a chemical change occurs: a. atoms are rearranged. b. the law of conservation of mass is always obeyed. c. the chemical properties of new substances are different from the ones you started with. d.

Chapter 7 Pretest: Chemical Reactions - School Specialty

C. The number of atoms in the products equals the number of atoms in the reactants. D. The states of matter of the products are independent of the states of reactants. 3. A chemical reaction has not occurred if the products have ____.

Chapter 7 Test: Chemical Equations - ProProfs Quiz

Unit 11: Chapter 7 - Chemical Reactions Review Answer Key. Define: Reactant: A substance that undergoes change in a chemical reaction. Product: New substances formed as a result of a chemical reaction. Catalyst: A substance that affects the rate of a chemical reaction without being used up in the reaction.

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